

DATE: March 2017

# MATERIAL SAFETY DATA SHEET (MSDS) OXYGEN

Version: 4

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Ref. No.: MS027

(Please ensure that this MSDS is received by the appropriate person)

### 1 PRODUCT AND COMPANY IDENTIFICATION

Product Name OXYGEN Chemical Formula O2

Trade Names Oxygen, Compressed

Oxygen, Instrument Grade (N2.5) Oxygen, EP Grade (N2.7) Oxygen, IG Zero (N4.5) Oxygen, UHP (N4.5) Medical Oxygen

Oxygen Agrigas Oxygen Econopack Oxygen Portapak

Colour coding Compressed, IG, EP, IG Zero & UHP

cylinders all have black bodies. Relevant decals or stencilling depict actual grades.

Medical Oxygen

Black Body with a white shoulder

Oxygen AgriGas

Black body with an orange valve guard

Oxygen Econopack

Black body with a blue valve guard Compressed, IG, EP, IG Zero & Medical grades have 3 SO- Brass, 5/8 inch BSP

right hand female valves.

Medical oxygen cylinders could also have the revenant Pin Index valves fitted. UHP grade has the Neriki-Brass 5/8 inch BSP right hand female valve fitted.

Company Identification African Oxygen Limited

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EMERGENCY NUMBER 0860 020202 or (011) 873 4382

(24 hours)

# **2 COMPOSITION/INFORMATION ON INGREDIENTS**

Chemical Name Oxygen
Chemical Family Oxidant
CAS No 7782-44-7
UN No 1072
ERG No 122

Hazchem Warning 5 A Non-flammable Gas

### 3 HAZARDS IDENTIFICATION

# **Main Hazards**

Valve

All cylinders are transportable gas containers. Oxygen is non-flammable, but readily supports combustion. Never permit oil, grease or other readily combustible substance to come into contact with high concentrations of Oxygen.

### Adverse Health Effects

Central nervous system toxicity including dizziness, convulsions and loss of consciousness can occur after only 2-3 hours of exposure to pure oxygen at 2 or more atmospheres. Retrosternal soreness, associated with coughing and breathing difficulties, made worse by smoking and exposure to cold air can occur after breathing pure oxygen at atmospheric pressure for several hours.

# **Chemical Hazards**

Oxygen is non-flammable, but strongly supports combustion (including some materials which do not normally burn in air). Since dry Oxygen is non-corrosive, most materials of construction are suitable. Avoid all flammable materials.

### **Biological Hazards**

No known effect.

# **Vapour Inhalation**

Pure oxygen is a local irritant to mucous membranes and, with extended continued exposure, can be destructive to lung tissue.

### **Label Elements**

Hazard Pictograms



### Signal Word: Danger

### **Precautionary Statements:**

P220: Keep/Store away from clothing/.../ combustible materials (manufacturer/supplier or the competent authority to

specify applicable ignition sources).

P244: Keep reduction valves free from grease and oil. P370+P376: In case of fire: stop leak if safe to do so.

P403: Store in well ventilated place.

**Hazard Statements:** 

H270: May cause or intensify fire; oxidser.

### 4 FIRST AID MEASURES

Eye/Skin Contact
Ingestion
Inhalation

No known effect.
(See Section 3 above)

Prompt medical attention is mandatory in all cases of overexposure to Oxygen. Rescue personnel should be cognisant of extreme fire hazard associated with oxygen-rich atmospheres. Conscious persons should be assisted to an uncontaminated area and inhale fresh air. They should be kept warm and quiet. Quick removal from the contaminated area is most important. The physician should be informed that the patient has experienced hyperoxia.

# 5 FIRE FIGHTING MEASURES

# **Extinguishing Media**

As Oxygen is non-flammable, but strongly supports combustion; the correct type of extinguishing should be used depending on the combustible material involved.

### **Specific Hazards**

Oxygen vigorously accelerates combustion. Materials that would not normally burn in air could combust vigorously in atmospheres having high concentrations of Oxygen.

### **Emergency Actions**

If possible, shut off the source of escaping Oxygen. Evacuate area. All cylinders should be removed from the vicinity of the fire. Cylinders that cannot be removed should be cooled with water from a safe distance. Cylinders which have been exposed to excessive heat should be clearly identified and returned to supplier.

CONTACT THE NEAREST AFROX BRANCH.

### **Protective Clothing**

Safety goggles, gloves and safety shoes should be worn when handling cylinders.

# **Environmental Precautions**

As the gas is heavier than air, pockets of Oxygen-enriched air could occur. These could lead to the fire spreading rapidly. If possible, ventilate the affected area.

# 6 ACCIDENTAL RELEASE MEASURES

### **Personal Precautions**

Although Oxygen is not itself combustible, it supports and accelerates combustion. Clothes and other materials, not normally considered flammable, will burn fiercely in the presence of Oxygen, and can be set alight by a single spark, or even hot cigarette ash.

### **Environmental Precautions**

Oxygen does not pose a hazard to the environment. Beware of Oxygen-enriched atmospheres coming into contact with readily combustible materials. If possible, ventilate the affected area.

# **Small Spills**

Shut off the source of excess Oxygen. Ventilate the area.

### Large Spills

Evacuate the area. Shut off the source of the spill if this can be done without risk. Ventilate the area using forced-draught if necessary.



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### 7 HANDLING AND STORAGE

Do not allow cylinders to slide or come into contact with sharp edges. Cylinders of Oxygen should not be stored near cylinders of acetylene or other combustible gases. Oxygen cylinders may be stacked horizontally provided that they are firmly secured at each end to

prevent rolling. Prevent dirt, grit of any sort, oil or any other lubricant from entering the cylinder valves, and store cylinders well clear of any corrosive influence, e.g. battery acid. Compliance with all relevant legislation is essential. Use a "first in – first out" inventory system to prevent full cylinders from being stored for excessive periods of time. Keep out of reach of children.

### 8 EXPOSURE CONTROLS/PERSONAL PROTECTION

### **Occupational Exposure Hazards**

Avoid exposure to oxygen-enriched atmospheres, as this could result in clothing becoming saturated by oxygen. On ignition the clothing could burn fiercely resulting in serious burns.

### **Engineering Control Measures**

Engineering control measures are preferred to reduce exposure to Oxygen-enriched atmospheres. General methods include forced-draught ventilation, separate from other exhaust ventilation systems. Ensure that sufficient fresh air enters at, or near, floor level.

#### **Personal Protection**

Safety goggles, gloves and shoes should be worn when handling cylinders.

Skin

No known effect.

#### 9 PHYSICAL AND CHEMICAL PROPERTIES

**PHYSICAL DATA** 

Chemical Symbol O2
Molecular Weight 32,00
Specific Volume @ 20°C & 101,325 kPa 755 ml/g

Boiling Point 101,325 kPa 90.18 °K; -183 °C; 181.4 °F Density, gas @ 101,325 kPa and 20°C 1,33 kg/m<sup>3</sup>

Density, gas @ 101,325 kPa and 20°C  $\,$  1,33 kg/m  $^3$  Relative density (Air = 1) @ 101,325 kPa 1,053 Solubility in Water @ 101.325 kPa @ 25 °C

(Partial Pressure of O<sub>2</sub>) @ O °C 4.889 cm<sup>3</sup> O<sub>2</sub>/100 cm<sup>3</sup> water

Colour None
Taste None
Odour None

# 10 STABILITY AND REACTIVITY

### Conditions to avoid

The build up of Oxygen-enriched atmospheres as, depending on temperature, oxygen reacts with all of the elements, excepting the

inert gases, to form oxides. These reactions can sometimes be

Chronic Toxicity

Carcinogenicity

Mutagenicity

Reproductive Hazards

No known effect

No known effect

No known effect

No known effect

(For further information see Section 3. Adverse Health effects)

### 12 ECOLOGICAL INFORMATION

Oxygen is heavier than air and care should be taken to avoid the formation of Oxygen-enriched pockets. It does not pose a hazard to the ecology.

### 13 DISPOSAL CONSIDERATIONS

### **Disposal Methods**

Small amounts may be blown to atmosphere under controlled conditions. Large amounts should only be handled by gas supplier.

# **Disposal of Packaging**

The disposal of containers must only be handled by the gas supplier.

# 14 TRANSPORT INFORMATION

### **ROAD TRANSPORTATION**

UN No 1072 ERG No 122

Hazchem warning 5A Non-flammable Gas





### **SEA TRANSPORTATION**

IMDG 1072

Class

Packaging group

Label Non-flammable Gas

AIR TRANSPORTATION

ICAO/IATA Code 1072

Class Non-flammable

Packaging group 2.2

Packaging instructions

Cargo 200 Passenger 200

Maximum quantity allowed

- Cargo 100kg - Passenger 75kg

# 15 REGULATORY INFORMATION

EEC Hazard class Non-flammable

National legislation OHSact and Regulations 85 of 1993.

Reference SANS 10234 and its supplement.

# **16 OTHER INFORMATION**

# Bibliography

Compressed Gas Association, Arlington, Virginia Handbook of Compressed Gases – 3<sup>rd</sup> Edition Matheson. Matheson Gas Data Book – 6<sup>th</sup> Edition SABS 0265 - Labelling of Dangerous Substances

### 17 EXCLUSION OF LIABILITY

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